

Discovery of 4-[(2*S*)-2-[[4-(4-Chlorophenoxy)phenoxy]methyl]-1-pyrrolidinyl]butanoic Acid (DG-051) as a Novel Leukotriene A4 Hydrolase Inhibitor of Leukotriene B4 Biosynthesis[†]

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Both in-house human genetic and literature data have converged on the identification of leukotriene 4 hydrolase (LTA₄H) as a key target for the treatment of cardiovascular disease. We combined fragment-based crystallography screening with an iterative medicinal chemistry effort to optimize inhibitors of LTA₄H. Ligand efficiency was followed throughout our structure–activity studies. As applied within the context of LTA₄H inhibitor design, the chemistry team was able to design a potent compound **20** (DG-051) ($K_d = 26$ nM) with high aqueous solubility (> 30 mg/mL) and high oral bioavailability (> 80% across species) that is currently undergoing clinical evaluation for the treatment of myocardial infarction and stroke. The structural biology–chemistry interaction described in this paper provides a sound alternative to conventional screening techniques. This is the first example of a gene-to-clinic paradigm enabled by a fragment-based drug discovery effort.

Introduction

We recently reported the identification of sequence variants in the FLAP^a and LTA₄H genes that were associated with enhanced capacity for the production of leukotriene B₄ (LTB₄) and increased risk for myocardial infarction (MI) and stroke.^{1,2} These conditions feature inflammatory damage to the arterial wall via the formation of atherosclerotic plaque followed by its rupture and consequent thrombosis.³ Inflammatory cell infiltration, LTB₄ biosynthetic enzymes, and LTB₄ production are increased in atherosclerotic plaque.^{4–6}

Literature evidence further suggests that leukotrienes mediate vascular and cellular immune responses in asthma, allergy, and inflammatory diseases.⁷ Leukotriene synthesis inhibitors and cysteinyl leukotriene receptor antagonists have shown efficacy in the clinic, resulting in a number of launched therapeutic products (e.g., (*N*-(1-benzo[*b*]thien-2-ylethyl)-*N*-hydroxyurea), zileuton (A-64077),⁸ 1-(((1*R*)-1-(3-((*E*)-2-(7-chloro-2-quinolinyl)vinyl)phenyl)-3-(2-(2-hydroxy-2-pro-

panyl)phenyl)propyl)sulfanyl)methyl)cyclopropyl)acetic acid, montelukast,⁹ and cyclopentyl(3-(2-methoxy-4-(((2-methyl-phenyl)sulfonyl)carbamoyl)benzyl)-1-methyl-1*H*-indol-5-yl)-carbamate, zafirlukast (ICI 204,219).¹⁰ LTA₄H is a key enzyme in the leukotriene pathway, which catalyzes the final and rate-determining step in the synthesis of LTB₄.¹¹ Earlier studies¹² of the potential role of LTA₄H in disease have focused on allergy,¹³ dermatitis,¹⁴ and arthritis¹⁵ with a variety of inhibitors¹⁶ reported in the literature (Figure 1).

Several fragment-based drug discovery approaches for screening low molecular weight (< 300 Da) compounds have been reported in the literature, and these include X-ray crystallography, nuclear magnetic resonance (NMR), surface plasmon resonance, differential thermal denaturation, fluorescence polarization, and other techniques.^{17–20} In our structure-based approach, we used X-ray crystallography for both identification of hits and their subsequent structure-based evolution into advanced molecules. In our effort to maintain drug-like characteristics of the optimized molecules, in addition to obtaining their bound pose and tracking improvement in enzyme inhibitory activity (IC₅₀) for the intended molecular target, we also followed indices such as BEI (binding efficiency index), SEI (surface efficiency index), and LE (ligand efficiency) that simultaneously track three crucial variables: potency, polar surface area, and molecular weight.^{21,22} On the basis of the analysis of over 120 marketed drugs, a range of optimal values (SEI = 18 ± 8.7, BEI = 28 ± 7.9, and LE ~ 0.4), for these indices have been provided as a guide.²³

Results and Discussion

Initially, crystals of LTA₄H were grown (Supporting Information for details) and soaked with our fragment library,

[†]Protein Data Bank coordinates have been deposited under the following accession codes: compound **2** bound to LTA₄H, 3FHE; compound **5** bound to LTA₄H, 3FH8; compound **8** bound to LTA₄H, 3FTZ; compound **9** bound to LTA₄H, 3FUL; compound **14** bound to LTA₄H, 3FH5; compound **20** bound to LTA₄H, 3FH7.

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^aAbbreviations: APA, aminopeptidase A; APP1, aminopeptidase P1; APN, aminopeptidase N; AUC, area-under the curve; BEI, binding efficiency index; BSA, bovine serum albumin; CVD, cardiovascular diseases; FLAP, 5-lipoxygenase activating protein inhibitor; FOL, fragments of life; LTA₄H, human leukotriene A4 hydrolase; HWB, human whole blood; SEI, surface efficiency index; LE, ligand efficiency; LRAP, leukocyte-derived arginine aminopeptidase; PILS, puromycin-insensitive leucyl-specific aminopeptidase; PK, pharmacokinetic.

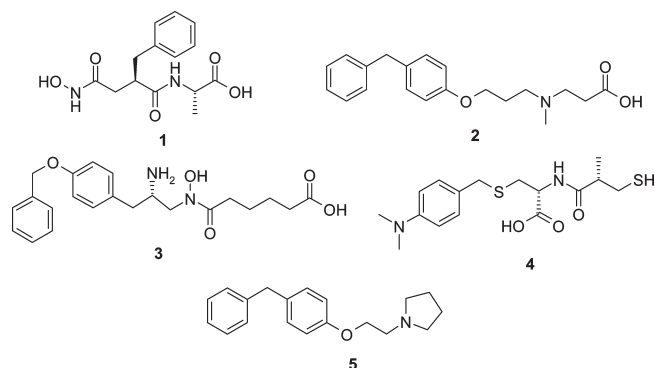


Figure 1. LTA₄H inhibitors reported in the literature.¹³

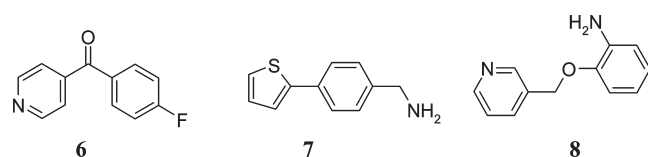


Figure 2. Structures of bound biaryl/heteroaryl FOL fragments.

termed “fragments of life” (FOL).²⁴ This customized set of ~1300 lead-like small molecules²⁰ is composed of natural metabolites, their derivatives, and heterocyclic derivatives that mimic protein-turn motifs (e.g., α -, β -, and γ -turns). These compounds have high solubility and chemical features allowing rapid synthetic elaboration. The FOL library, clustered into 6–8 diverse fragments per pool, was soaked with LTA₄H crystals and hits were reconfirmed by X-ray crystallography with a single fragment. This expedited approach provided several diverse biaryl/heteroaryl fragments bound to LTA₄H as chemical leads for further elaboration. The structures of three fragments are shown in Figure 2; their binding modes at the hydrophobic L-shaped active site pocket are shown in Figure 3. For each of the identified fragments, we observed both hydrophobic packing as well as specific hydrogen bond interactions with either crystallographic water (ex., ligand **6**, N–O (water) = 3.2 Å), active site residues (ex. ligand **8** with CO of Pro374), or water mediated interactions with active site residues (ex., ligand **7** with Gln136). The observed inhibition of LTA₄H hydrolysis (IC₅₀ = 5.38 mM, 98 μ M, and > 2 mM and for fragments **6**, **7**, and **8**, respectively) by these fragments prompted their further elaboration. The enzyme assay used recombinant human LTA₄H with LTA₄ as the substrate to determine IC₅₀ values (see Experimental Section, Biological Assays). In several instances, we found an acetate anion recruited from the crystallization media to bind to the catalytic zinc in the active site of LTA₄H. An acetate bound to zinc was observed in the absence (not shown) or presence of a bound fragment in the hydrophobic pocket (Figure 3). This mapping of two distinct binding sites by the biaryl fragments and an acetate anion provided our initial hypothesis for the overall trajectory of an LTA₄H inhibitor to be pursued by structure-based design. The SEI/BEI for fragments **6** and **7** were 7.6/11.3 and 7.4/21.1, respectively. A combination of factors including active site fit, synthetic feasibility, and ligand efficiency prompted us to select **6** for further optimization.

In the next series of experiments, we continued with our iterative chemistry–structural biology studies. Our focus was to (i) capture additional interactions in the hydrophilic pocket by chemical elaboration of the biaryl fragments and (ii)

evaluate linking of the two identified fragments, biaryl and acetate. Considering both literature evidence²⁵ and our structural insight, we endowed fragment **6** with an *N*-pyrrolidine ethanol moiety (Figure 4) to result in analogue **9**. In our hands, the binding mode of **9** to LTA₄H (Supporting Information Figure SF1b) was consistent with the observed pose for **6**. However, analogue **9** contained two potentially disruptive interactions with the enzyme. These were between the carbonyls of **9** and Trp311 and between the pyridinyl nitrogen of **9** and the oxygen of water WAT-33, which were in close proximity (ligand CO–CO (Trp311) = 3.0 Å and N–O (WAT-33) = 3.12 Å). Thus, we observed only limited improvement in LTA₄H inhibition for **9** vs **6** (IC₅₀ = 199 and 5380 μ M, respectively). These limitations were addressed in analogue **5**, which did not feature any of these repulsive interactions (Supporting Information Figure SF1c) and consequently showed potent enzyme inhibition (IC₅₀ = 180 nM). The binding mode of compound **2** (Supporting Information Figure SF1d) provided further insight into chemical design. X-ray crystallographic data showed good complementarity between the biaryl moiety and the hydrophobic surface of the active site channel. We also observed an ionic interaction of the carboxylate group with the Zn ion (Zn²⁺–COO[−] distance = 1.49 Å). This –COO–Zn contact is contrary to previously published data based on extended X-ray absorption fine structure (EXAFS) spectroscopy of LTA₄H, which concluded that there was a lack of interaction of carboxylate containing inhibitors with the zinc atom of the enzyme.²⁶ As anticipated, molecule **5** also featured sound ligand efficiency (0.44), although it displayed very high SEI value (54.1).

Combining the structural evidence accumulated thus far, we explored a series of functional groups to tether the identified biaryl fragments to the carboxylate. A number of linkers were considered and evaluated, with a goal for the optimized linker to capture additional interactions with LTA₄H active site residues. Representative examples of these linkers are shown in Supporting Information Figure SF2: (a) the pyridine analogue **10** is based on potentials for capturing π – π , π -to-edge or π -cation interactions with Tyr267 and (b) replacement of the ether oxygen of **5** with tertiary nitrogen, yielding piperazine analogues **11** and **12**, which were designed to fill the available volume and to mediate interaction with Gln134/Gln136. Structures of analogues **10**–**12** are shown in Figure 4. Analogue **10** showed relatively weak inhibitory activity (IC₅₀ = 8.7 μ M), while derivatives **11** and **12** displayed better inhibitory potency (IC₅₀ = 1.4 and 0.340 μ M, respectively). The bound poses of the pyridine analogue **10** and the iodo-analogue **12** are shown in Supporting Information, parts a and b of Figure SF3, respectively.

Despite a good fit into the binding pocket, moieties featuring a tertiary amine group in compounds **2** and **5** showed no direct interaction with Gln134 and Gln136 (Supporting Information Figure SF1c,d). In a subsequent chemical modification, we replaced the aminoethylether chain of **5** with either (*S*) or (*R*) prolinol to result in molecules **13** and **14** (Figure 5a). Both enantiomeric derivatives, (*S*)-isomer **13** and (*R*)-isomer **14**, provided improved activity compared to analogue **5** (IC₅₀ = 244 and 87 nM, respectively). Moreover, the pyrrolidine nitrogen of **13** and **14** is positioned to exert strong interactions with Gln134 (2.76 Å) and Gln136 (2.83 Å) (Figure 5b,c), as evidenced from the structural studies. The side chain of Gln134 underwent a ligand induced conformational change from its native position in the enzyme²⁷ to accommodate the interactions with **14**. This side chain

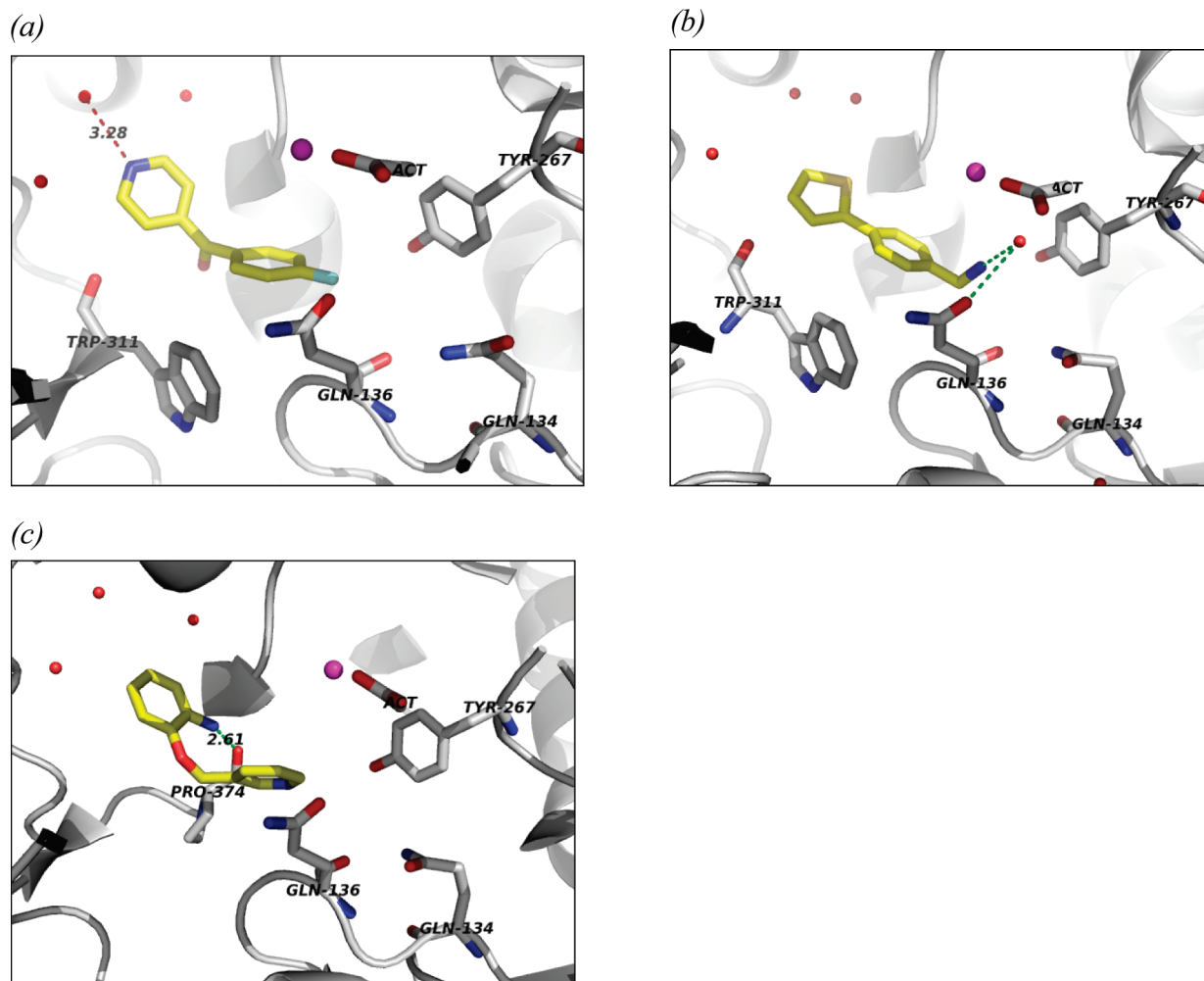


Figure 3. Bound pose of biaryl fragments at the hydrophobic pocket and acetate bound to catalytic zinc. (a) Fragment 6 with N–O (water) = 3.2 Å and Zn–OCOCH₃ = 1.98 Å, resolution 1.8 Å; (b) fragment 7 and acetate with Zn–OCOCH₃ = 1.98 Å, resolution 2.0 Å; (c) fragment 7 and acetate with Zn–OCOCH₃ = 2.19 Å, resolution 2.05 Å. In each figure, acetate (labeled as ACT) is bound to active site zinc. The catalytic Zn ion is shown in magenta and water molecules are shown as red spheres.

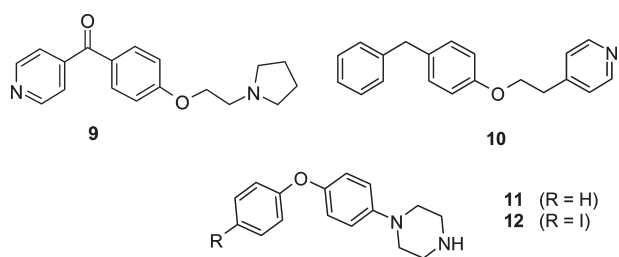


Figure 4. Structures of analogues 9–12.

rotamer led to the formation of a new hydrogen bond between the amide nitrogen of Gln134 and the peptidyl oxygens of Gln136 (2.82 Å) and Gln134 (intraresidue hydrogen bond, 3.04 Å) (Figure 5b). The X-ray cocrystal structure of the corresponding (*S*)-enantiomer **13** revealed a weaker interaction with Gln136 (3.35 Å), whereas Gln134 retained its original orientation. In support of the observed 3-fold difference in potency, cocrystal structures of **13** lacked the hydrogen bond network observed for its enantiomer **14** (Figure 5c). For both ligands, the methyleneoxy chain between the biphenyl group and the pyrrolidine ring afforded optimal positioning of the ligand within the LTA₄H binding pocket (Figure 5b,c). Of the evaluated linkers, the prolinol-derived ligand impacted

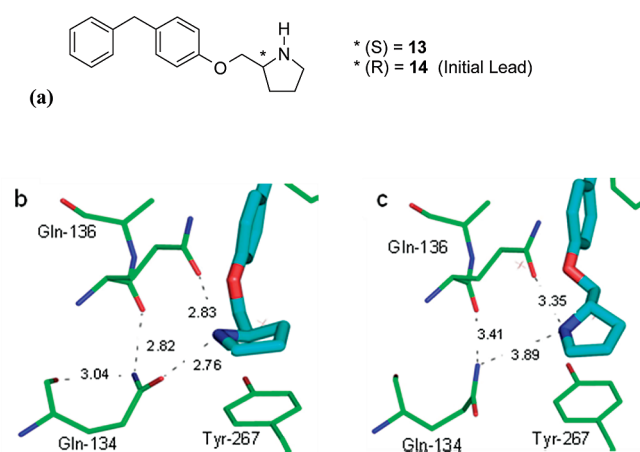


Figure 5. (a) Shows the chemical structures for the (*S*) and (*R*) prolinol-derived analogues **13** and **14**, respectively, and (b) and (c) shows bound orientations of (*S*) and (*R*) prolinol-derived analogues **13** and **14**, respectively. (b) Hydrogen bond network involving Gln134 and Gln136 in the complex with **14**. (c) Identical view for the corresponding (*S*) enantiomer, **13**. Note: in the bound poses of these two enantiomers, Gln134 has different orientation, with orientation of Gln134 with **14** being identical to that in the apoenzyme (not shown).

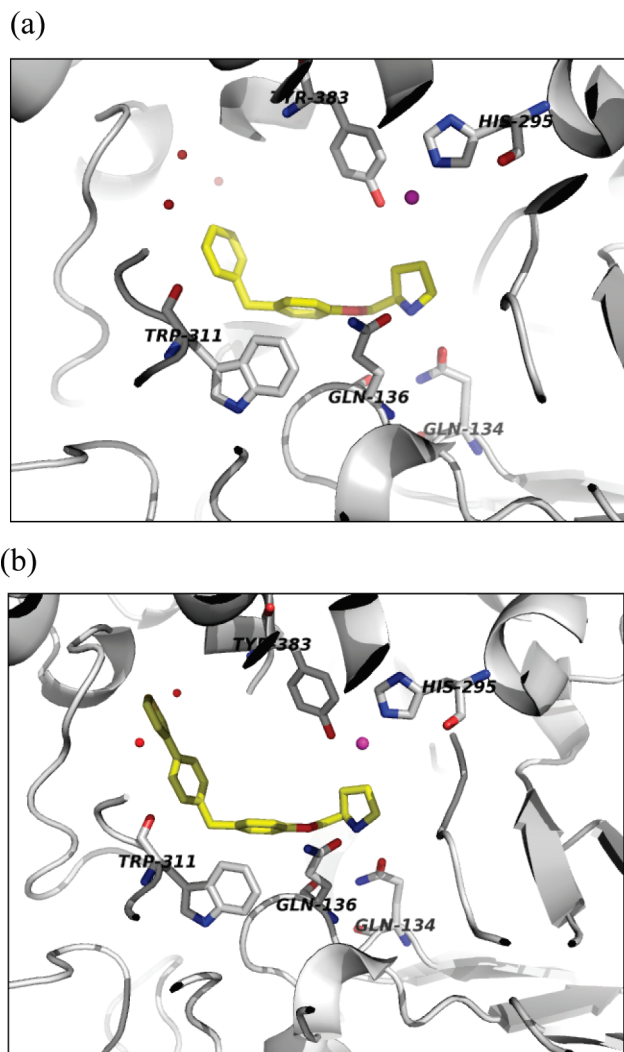


Figure 6. Displacement of “structural” water molecule with *para*-3'-thienyl group of ligand **15**. (a) X-ray cocrystal structure of LTA₄H in complex with **14** at 1.63 Å resolution showing three crystallographic waters in the hydrophobic pocket and (b) X-ray cocrystal structure of **15** at 1.69 Å resolution showing only two crystallographic water molecules in the hydrophobic pocket. The thiophene has displaced the center water molecule present in (a). The catalytic Zn ion is shown in magenta and water molecules are shown as red spheres.

the Gln134 side chain movement the most, securing it as a tether of choice for subsequent ligand elaboration.

We attempted further optimization by varying the nature of the biaryl group. Considering the hydrophobic nature of the pocket occupied by this pharmacophore, we incorporated lipophilic substituents onto the terminal phenyl ring. Substitution at the *para*-position was projected to displace one or more of the three conserved water molecules deep within the active site pocket (Figure 6a), thus potentially improving affinity of the molecule toward LTA₄H and blocking the anticipated cytochrome P450 (CYP) mediated metabolism at this site. We found that relatively large groups including thiophenes and thiazoles did in fact displace the central water molecule (WAT-762) as in the cocrystal structure of **14** (Figure 6b), and these modifications provided potent LTA₄H inhibitors. Compound **15** showed a 21-fold improvement in enzyme inhibitory activity (IC_{50} = 4 nM) vs the unsubstituted analogue **14** (IC_{50} = 87 nM). Unfortunately, these derivatives

Table 1. Biological and Pharmacological Data for Selected Compounds

compd	14	17	20		
IC_{50} (nM) (enzyme)	87	30	47		
K_d (nM)	ND	ND	25		
IC_{50} (nM) (HWB) [†]	449	533	37		
pK_a ^a	10.41	10.39	4.02, 8.35 (pI = 6.18) ^b		
microsome stability ^c	10.5	86.6	95.2		
PK parameters ^d					
species	rat	mouse	rat	dog	monkey
dose (mg/kg)	10	10	10	10	10
$t_{1/2}$ (h) ^e	0.67	3.59	2.57	4.02	5.54
C_{max} (μ M) ^f	1.46 (iv)	1.02	4.8	19.3	18.9
AUC (h·ng/mL)	471 (iv)	2362	12968	56045	52410
% F ^g	ND	85	87	252	186
V_d (mL/kg) (iv) ^h	6121	6324	2351	1564	2073
Cl (mL/min/kg) (iv) ⁱ	71.0	61.0	11.0	7.6	6.0

^a pK_a values are calculated using ACD laboratories software for **14** and **17**, and experimental values are reported for compound **20** (pK_{a1} = 4.02, pK_{a2} = 8.35). ^b Isoelectric constant. ^c Microsome stability indicates parent remaining after 30 min incubation in human liver microsomes. ^d Unless otherwise indicated, pharmacokinetic (PK) values are for oral administration (po). ^e The plasma elimination half-life, $t_{1/2}$, was determined by linear regression of the terminal phase of the logarithm plasma concentration–time curve. ^f The maximum plasma concentration (C_{max}) is the observed peak plasma concentration after an oral dose. ^g Oral bioavailability ($F\%$) was determined using AUC (area under the plasma concentration–time curve from time zero to infinity normalized for dose) in po, (except when indicated for iv) ^h Volume of distribution. ⁱ Predicted clearance. ND = not determined.

were found to be cytotoxic and relatively nonspecific in a broad panel screen (Cerep) for off-target activity.²⁸ Replacement of the heterocycle with a *p*-chlorine substituent to yield **16** addressed the issue of cellular toxicity. As shown by structural studies, the *para*-Cl group did not displace the bound water molecules. Values for the three ligand efficiency indices were: LE = 0.21 for compound **6** and 0.24 for compound **10** vs LE = 0.49 for compound **14**, and SEI and BEI = 7.6 and 11.3 for compound **6** vs 33.1 and 26.4 for compound **14**, respectively. Thus, the structure-guided elaboration of fragment **6** to compound **14** was on track for simultaneously enhancing key ligand parameters (Supporting Information Table ST1 and Figure SF4).

As reported above, we found that **14** inhibited LTA₄H with an IC_{50} of 87 nM. In a physiologically relevant secondary assay, human whole blood (HWB) was used to assess inhibition (IC_{50}) of LTB₄ production after stimulation with Ca²⁺ ionophore (see Experimental Section, Biological Assays). In this assay, **14** showed an IC_{50} of 449 nM, a 5-fold decrease compared to the in vitro data. The cocrystal structure of **14** bound to the enzyme is shown in Figure 6b.

In an attempt to further optimize compound **14**, we replaced the methylene bridge with an ether moiety providing compound **17** with a 2-fold increase in enzyme inhibitory potency (IC_{50} = 49 nM) and HWB assay (IC_{50} = 284 nM). The X-ray analysis confirmed an essentially identical binding mode for **17**. This modification also afforded improvement in key PK parameters, such as half-life and systemic exposure (for **14** $t_{1/2}$ = 0.67 h; AUC_{0-1hr} = 471 h·ng/mL and for **17**: $t_{1/2}$ = 3.6 h; AUC_{1hr} = 2362 h·ng/mL), presumably due to eliminating metabolism at the methylene site (Table 1).

Focusing on the second binding element revealed by fragment screening, the acetate ion bound to the catalytic zinc (Figure 3), we optimized the carboxylic acid pharmacophore by introducing carbon chains of variable length between the

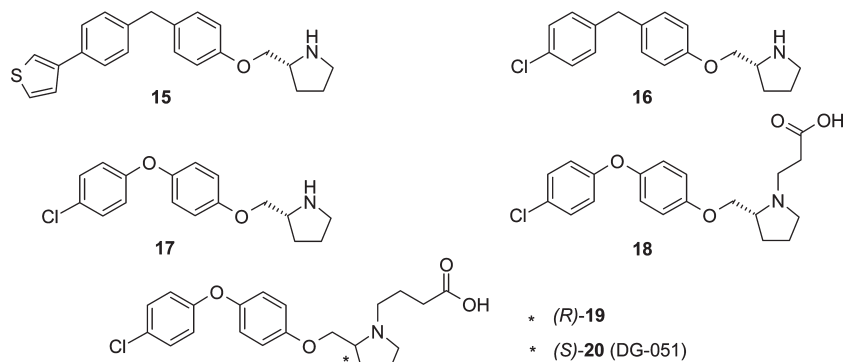


Figure 7. Advanced intermediate and the clinical candidate compound.

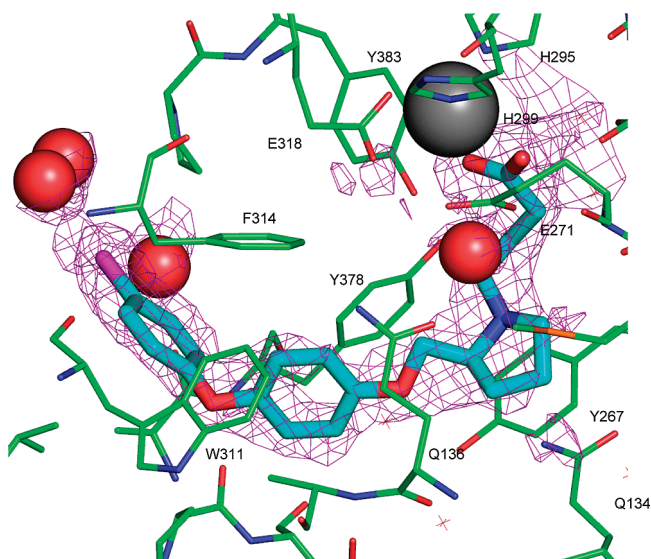


Figure 8. X-ray cocrystal structure of **20** in complex with LTA₄H showing the displacement of two water molecules located near Zn by the carboxylic acid group. The $F_o - F_c$ electron density map is contoured at 2σ .

pyrrolidine nitrogen and the carboxylate. Both two- and three-carbon chains containing carboxylic acid derivatives (**18** and **19**, Figure 7) showed comparable activities in the enzyme assay and were potent in the HWB assay with IC_{50} values of 23 and 33 nM, respectively. A two-carbon linker previously was reported in compound **2** to undergo metabolic *N*-oxidation followed by Cope reaction to furnish acrylic acid and *N*-hydroxyl amine leading to toxicity in vivo.²⁹ On the basis of our concern that **18** may undergo similar metabolic conversion, the butyric acid derivative **19** featuring a three-carbon linker was selected for further evaluation.

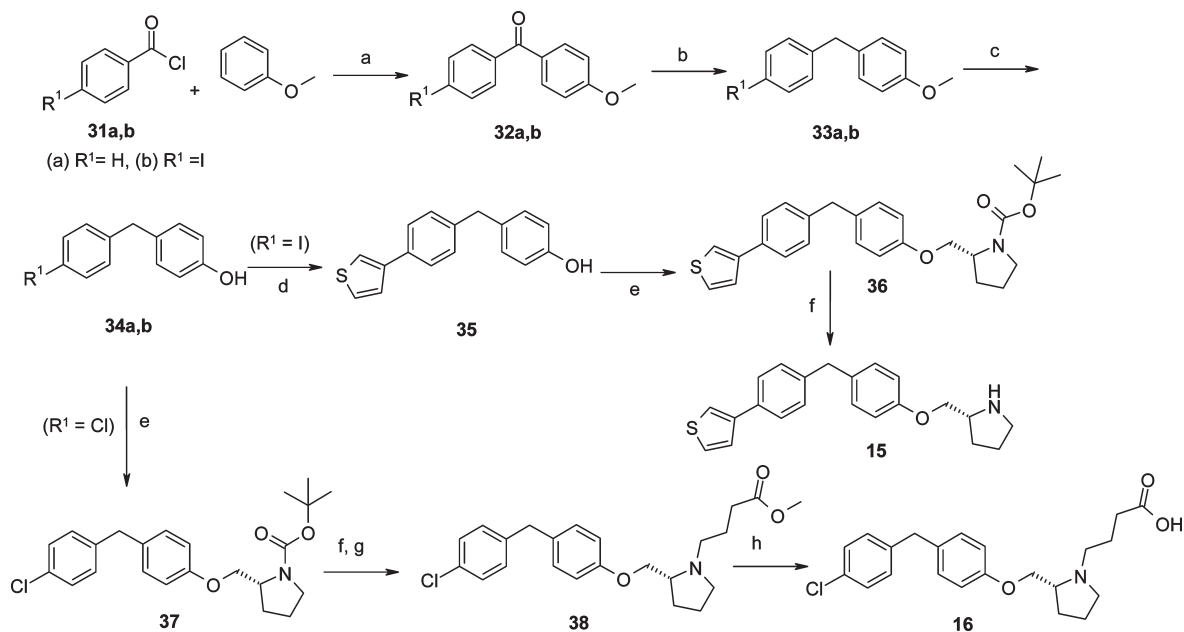
Given the relatively high cost of *D*-proline, the corresponding (*S*)-enantiomer derived from *L*-proline was prepared for comparison (**19** and **20**, respectively). These molecules were equipotent in the HWB assay ($IC_{50} = 33$ and 37 nM, respectively). Therefore, the synthetically more feasible (*S*)-enantiomer **20** was identified as the lead compound for further evaluation. The X-ray crystal structure of **20** bound to LTA₄H showed that the carboxylate displaced two of the three water molecules near Zn²⁺ by chelating to the metal. One oxygen of the carboxylate group was a direct ligand of Zn²⁺ adopting a slightly distorted tetrahedral coordination geometry with His295, His299, and Glu318, while the second oxygen interacts with Glu296 and the remaining water molecule

(WAT-545 in the **20** structure) of the solvent cluster (Figure 8). Compound **20** was a potent inhibitor in the enzyme assay with an $IC_{50} = 47$ nM. Although the molecule gained binding affinity through interaction with Zn²⁺, in contrast to **14** it did not participate in hydrogen-bond interactions with Gln136 or Gln134, although the tertiary nitrogen (of **20**) appears to show water mediated interaction with Glu271. Inhibitor **20** provided over 4-log improvement in activity compared with the biaryl compound **6** that was identified through crystallographic screening of our in-house fragment library.

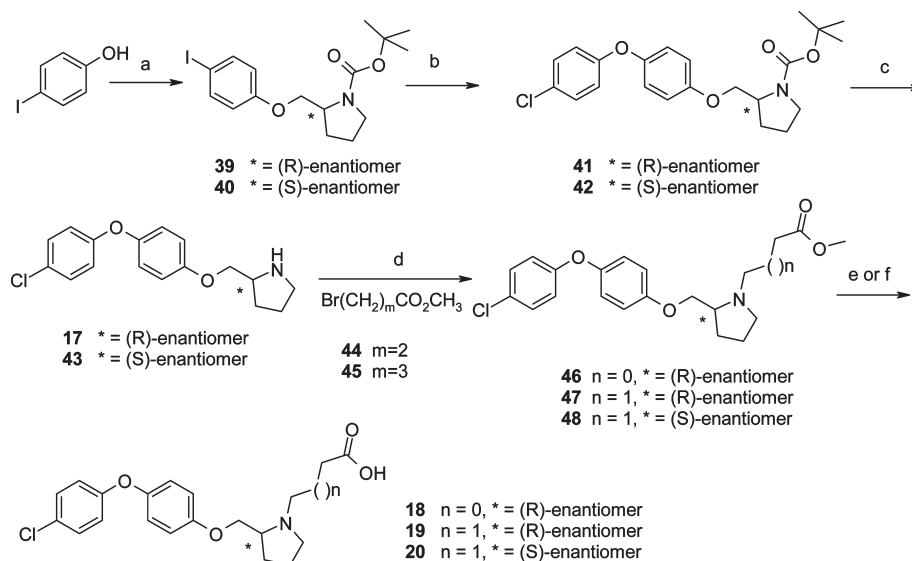
Although useful for screening and rank ordering of compounds for inhibitory potency, the *in vitro* enzyme assay was formatted as an end-point assay. It required relatively high concentrations of both the enzyme and bovine serum albumin (BSA) in the reaction mixture (see Experimental Section, *In Vitro* Enzyme Assay). Therefore, we used isothermal microcalorimetry to determine the equilibrium dissociation constant, K_d , of **20** binding to LTA₄H. The measured value of 25 nM compared favorably against both enzymatic and whole blood assay data, suggesting lack of significant plasma binding and high cellular permeability for the molecule.

The introduction of the carboxylic acid group into **20** allowed for the optimization of other compound properties. Specifically, aqueous solubility at the isoelectric point (pH ~ 6.18) was high (> 30 mg/mL), streamlining formulation and oral administration protocols for **20**. The molecule showed good absorption and low clearance across all species tested with very high oral bioavailability (Table 1). The half-life in dog and monkey predicted a similar value in human, sufficient to allow once daily dosing. Furthermore, for compound **20**, we were able to lower the volume of distribution ($V_d \sim 2000$ mL/kg) compared to early compounds (**14** and **17**), resulting in higher plasma concentration. Considering our target indication, it is beneficial to have more drug in the systemic circulation than in the tissues, as blood cells and the vessel wall are the main target compartments in cardiovascular diseases. Very high values of C_{max} and AUC based on dose across all species indicated the potential for good *in vivo* potency. It is worth noting that compound **20** both displayed excellent PK profile across various species as well as good physicochemical properties. This molecule also featured excellent ligand efficiency [(LE = 0.38), BEI (18.8), and SEI (12.4)], values similar to several successful marketed drugs.²⁰ Thus, crystallography driven medicinal chemistry optimization provided compound **20** with enhanced potency against its molecular target (LTA₄H).

Off-Target Activity Profiling. The activity and selectivity of **20** was assessed in a variety of *in vitro* assays. LTA₄H is a

Scheme 4^a

^a (a) AlCl_3 , nitrobenzene, 90%; (b) Et_3SiH , TFA, 99%; (c) BBr_3 , CH_2Cl_2 ; (d) 3-thiophene-boronic acid, K_2CO_3 , Pd/C, $i\text{PrOH}/\text{H}_2\text{O}$, 99%; (e) NaH, DMF, (30); (f) 4 N HCl/*p*-dioxane, 53% (from 35 to 15); (g) K_2CO_3 , DMF, $\text{Br}(\text{CH}_2)_3\text{CO}_2\text{CH}_3$, rt, 35%; (h) NaOH, MeOH, 87%.

Scheme 5^a

^a (a) KO-*t*Bu, DMF, (29) or (30), 92%; (b) *N,N*-dimethylglycine, Cs_2CO_3 , Cu(I), 4-chlorophenol, 56%; (c) 4 N HCl, *p*-dioxane, 83%; (d) K_2CO_3 , DMF, (e) aq NaOH, MeOH, 36%; (f) HCl/ Et_2O , 93%; (g) (46), 1:1 conc HCl/*p*-dioxane, 60 °C, 4 h, 68%.

alkylation of the secondary amine with methyl-4-bromobutyrate and subsequent base hydrolysis of the ester (38), provided the analogue 16 (Scheme 4).

The general synthesis of the diphenyl ether derivatives 17–20 is shown in Scheme 5. Reaction of 4-iodophenol with BOC-tosylates 29 or 30, using Cu(I) mediated Ullmann coupling³⁵ of the iodo-phenyl intermediate 39/40 with 4-chlorophenol, provided the corresponding diphenyl ether derivatives 41/42. Deprotection of the *N*-BOC derivatives under acidic conditions provided the amine derivatives 17 ((*R*)-enantiomer) and 43 ((*S*)-enantiomer) as hydrochloride salts. Alkylation of the secondary amine (17 or 43) with a 4-bromobutyrate (45) or with 3-bromopropionate under

basic conditions provided the corresponding esters (46–48). The hydrolysis of the esters followed by acidification or under acid hydrolysis conditions provided the desired analogues 19–20.

Conclusion

In summary, the development of 20 illustrates a number of trends in modern drug discovery. These include (a) the use of human population genetic studies to identify new therapeutic targets for major common diseases, (b) the utility of a fragment-based screening for the identification of novel chemotypes empowering optimization studies, and (c) iterative structural biology–medicinal chemistry process leading to

accelerated lead optimization. As a result of the structure-assisted SAR and simultaneous optimization of pharmacokinetics, pharmacodynamics, selectivity, and solubility along with the minimized potential for toxicity, we have identified compound **20** as the clinical candidate. Specifically, our genetic studies linked LTA₄H genes that were associated with increased risk for myocardial infarction (MI) and stroke.^{1,2} To the best of our knowledge, this is the first example of a gene-to-clinic paradigm enabled by the iterative use of fragment-based drug discovery and structure-guided medicinal chemistry, providing a candidate optimized for ligand parameters and imparting desirable drug characteristics. DG-051 recently entered human phase II clinical trials in CVD.

Experimental Section

Chemistry. All reagents, starting materials, and anhydrous solvents were obtained from commercial sources and used without further purification unless otherwise noted. Concentration refers to evaporation under vacuum using a Büchi rotatory evaporator. NMR spectra were recorded at 400 or 500 MHz (Varian Instruments) in the solvent indicated, and TMS was used as an internal reference. ACDLabs NMR software was used to process FIDs to generate spectral parameters (ppm/Hz). Coupling constants (*J*) are given in Hz. Mass spectra were obtained using either APCI or electrospray ionization (PE-SCIEX single-quad or Agilent mixed-mode units). Elemental analyses were carried out by Galbraith Laboratories, Inc. (Knoxville, TN) or Midwest Microlab, LLC (Indianapolis, IN). Compounds were purified by flash chromatography using silica gel (MP EcoChrom, 32–63D, 60 Å). Typically, 1:10 to 1:20 ratio of the crude product to silica gel was used. Purity of all final products was determined by analytical HPLC and/or by LC-MS to be ≥95%. HPLC purity of compounds were measured with a reversed-phase HPLC (Phenomenex Prodigy C₁₈ column, 4.6 mm × 150 mm, 5 μm, typically at 254 nm, unless indicated) with one of the following two conditions. HPLC-1: compounds were eluted using a gradient of 90/10 to 10/90 A/B over 40 min at a flow rate of 1.0 mL/min, where solvent A was aqueous 0.05% TFA and solvent B was CH₃CN (0.05% TFA). HPLC-2: compounds were eluted using a gradient of 95/5 to 5/95 A/B over 40 min at a flow rate of 1.0 mL/min, where solvent A was aqueous 0.05% TFA and solvent B was CH₃CN (0.05% TFA). For HPLC data, (final products), peak area percent and retention time (*t_R* in minutes) are provided.

4-[2-(4-Benzylphenoxy)ethyl]pyridine (10). The 4-pyridine ethanol (123 mg, 1.0 mmol) was dissolved in THF (3 mL) and solutions of 4-benzylphenol (240 mg, 1.3 mmol) in THF (1 mL) and Ph₃P (315 mg, 1.2 mmol) in THF (1 mL) were added. The reaction was cooled to 0 °C, and a solution of diisopropylidiazodicarboxylate (242 mg, 1.2 mmol) in THF (1 mL) was added, followed by stirring at rt for 4 h. The reaction was heated to 70 °C and maintained for 16 h. The solvent was removed in vacuo, and the resulting material was purified by silica gel flash chromatography (20–50% EtOAc/hexanes, gradient) and stirred with 2 M HCl in Et₂O (excess) at rt to give the compound **10** as the hydrochloride salt (0.89 g, 27%). MS *m/z*: 290 (M + H). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 8.80 (d, *J* = 6.8 Hz, 2H), 7.96 (d, *J* = 6.8 Hz, 2H), 7.26 (m, 2H), 7.17 (m, 3H), 7.12 (d, *J* = 8.8 Hz, 2H), 6.84 (d, *J* = 8.8 Hz, 2H), 4.30 (t, *J* = 6 Hz, 2H), 3.86 (s, 2H), 3.31 (t, *J* = 6.3 Hz, 2H). HPLC-1: 99.7%, *t_R* = 19.5 min.

1-(4-Phenoxyphenyl)piperazine Hydrochloride (11). A solution of 4N HCl in dioxane (30 mL) and **10b** (3.0 g, 8.46 mmol) were stirred at rt for 1 h. EtOAc (30 mL) was added and solid was collected by filtration and washed with EtOAc (2 × 10 mL) and the solid was dried under vacuo to provide the title compound **11** (2.71 g, 98.2%) as a white solid. MS *m/z*: 255 (M + H). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 9.11 (br s, 1 H) 7.65

(d, *J* = 8.9 Hz, 2 H) 6.95–7.07 (m, 4 H) 6.75 (d, *J* = 8.9 Hz, 2 H) 3.30–3.36 (m, 4 H) 3.22 (br s, 4 H).

1-[4-(4-Iodophenoxy)phenyl]piperazine Hydrochloride (12). A solution of 4N HCl in dioxane (15 mL) and **12b** (1.1 g, 2.3 mmol) were stirred at rt for 1 h. EtOAc (30 mL) and hexanes (30 mL) were added and solid was collected by filtration and washed with EtOAc (2 × 15 mL) and the solid was dried under vacuo to provide the title compound **12** (1.11 g, 95%) as an off-white solid. MS *m/z*: 381 (M + H). LCMS: 98%. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 9.29 (br. s., 1 H) 7.27–7.41 (m, 2 H) 7.01–7.11 (m, 3 H) 6.89–6.99 (m, 4 H) 3.30–3.36 (m, 4 H) 3.18–3.25 (m, 4 H).

(2S)-2-[4-Benzylphenoxy)methyl]pyrrolidine (13). To a solution of 4-benzylphenol, **21** (0.10 g, 0.56 mmol) in anhydrous DMF (1 mL) at 0 °C was added a 60% dispersion of NaH in mineral oil (0.02 g, 0.75 mmol) portionwise over 5 min. The resulting slurry was stirred at 0 °C for 10 min before a solution of (*S*)-tosylate **29** (0.20 g, 0.56 mmol) in DMF (2 mL) was added dropwise over 5 min. The mixture was stirred at 95 °C for 8 h and then stirred at rt for 16 h. The reaction mixture was poured over ice and then concentrated under reduced pressure. The residue was extracted into EtOAc (20 mL) and sequentially washed with water (10 mL), saturated aqueous NaHCO₃ (2 × 10 mL), water (2 × 10 mL), and brine (2 × 10 mL). The combined organic layer was dried over anhydrous Na₂SO₄, filtered, and the solvent was removed in vacuo. The residue was chromatographed on silica gel (10 g) and eluted with 30% EtOAc/hexanes to afford (*S*)-2-(4-benzylphenoxy)methyl-1-BOC-pyrrolidine (0.15 g, 73%) as a tan oil. To a solution of this product (0.12 g, 3.26 mmol) was added 4 M HCl in dioxane (15 mL) at rt. The resulting mixture was stirred overnight. The solvent was removed in vacuo to obtain an off-white solid. The solid was triturated with Et₂O to provide compound **13** (55 mg, 63%) as a white solid. MS *m/z*: 268 (M + H). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 7.24–7.33 (m, 2H), 7.13–7.23 (m, 5H), 6.91 (d, *J* = 8.6 Hz, 2H), 4.16–4.26 (m, 1H), 4.09 (dd, *J* = 8.3, 10.47 Hz, 1H), 3.80–3.94 (m, 3H), 3.12–3.25 (m, 2H), 2.05–2.17 (m, 1H), 1.82–2.03 (m, 2H), 1.63–1.80 (m, 1H). HPLC-1: 99.4%, *t_R* = 15.1 min.

(R)-2-(4-Benzylphenoxy)methylpyrrolidine (14). Similar to the preparation of compound **13**, the (*R*)-BOC-intermediate was prepared from 4-benzylphenol (103 mg, 0.56 mmol), NaH (19 mg, 0.75 mmol) in 1 mL DMF and (*R*)-tosylate **30** (200 mg, 0.56 mmol) in DMF (3 mL) to provide after aqueous workup a crude product which following purification by silica gel (10 g) using 30% EtOAc/hexanes as eluent provided the BOC-protected intermediate (150 mg, 73%) as a solid. MS *m/z*: 368 (M + H). ¹H NMR (400 MHz, CDCl₃) δ: 1.46 (m, 9H), 1.84–2.03 (m, 4H), 3.39 (m, 2H), 3.73–3.93 (m, 3H), 4.10 (m, 2H), 6.84 (m, 2H), 7.08 (m, 2H), 7.17 (m, 2H), 7.26 (m, 3H). HPLC-1: 99.8% To a solution of the BOC-protected intermediate (0.12 g) in dioxane (2 mL) was added 4N HCl in dioxane (8 mL) at rt and the resulting mixture was stirred for 1 h at rt. The solvent was removed in vacuo to obtain a thick oil. The oil was triturated with Et₂O to obtain a white solid (2.5 g). The solid was recrystallized with toluene (20 mL) to obtain the title product (58 g, 53%) as a crystalline solid (**14**). MS: *m/z* 268 (M + H). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 7.15–7.20 (m, 5H), 7.27 (m, 2H), 6.91 (d, 2H, *J* = 6.8 Hz), 4.19 (dd, 1H, *J*₁ = 8.8, *J*₂ = 3.2 Hz), 4.13 (dd, 1H, *J*₁ = 7.2 Hz, *J*₂ = 6.4 Hz), 3.88 (s, 2H), 3.87 (m, 1H), 3.18 (m, 2H), 2.09 (m, 1H), 1.98 (m, 1H), 1.89 (m, 1H), 1.73 (m, 1H). HPLC-1: 98.8%, *t_R* = 15.2 min.

(R)-2-[4-(4-Thiophen-3-yl-benzyl)phenoxy)methyl]pyrrolidine (15). To an argon purged solution of **35** (9.0 g, 34 mmol) and (*R*)-2-(toluene-4-sulfonyloxymethyl)pyrrolidine-1-carboxylic acid *tert*-butyl ester (12.6 g, 36 mmol) in DMF at 0 °C was added NaH (60% dispersion in mineral oil, 1.62 g, 40 mmol). The reaction was stirred at 0 °C for 10 min and warmed to 85 °C for 16 h. The reaction mixture was poured into aqueous NaHCO₃ (500 mL) and extracted with EtOAc (3 × 150 mL). The organic

layer was washed with water (3 × 50 mL) and dried over anhydrous MgSO₄. The volatiles were removed and the residue was stirred with 4N HCl in dioxane (40 mL) for 16 h. The mixture was diluted with THF (450 mL). The resulting solids were filtered and rinsed with Et₂O to give the compound **15** (7.0 g, 53%). MS *m/z*: 351 (M + H). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 7.79 (d, *J* = 1.61 Hz, 1H), 7.58–7.66 (m, 3H), 7.51 (dd, *J* = 1.0, 5.0 Hz, 1H), 7.15–7.27 (m, 4H), 6.92 (d, *J* = 8.5 Hz, 2H), 4.20 (d, *J* = 3.6 Hz, 1H), 4.08 (s, 1H), 3.90 (s, 3H), 3.12–3.25 (m, 2H), 2.05–2.19 (m, 1H), 1.82–2.04 (m, 2H), 1.64–1.80 (m, 1H). HPLC-1: 99.9%, *t*_R = 21.0 min. Anal. calcd (C₂₂H₂₃NOS): C, H, N.

4-[(*R*)-2-[4-(4-Chlorobenzyl)phenoxy]methyl]pyrrolidin-1-yl]-butyric Acid (16**). The methyl ester **38** (0.5 g, 1.25 mmol) was dissolved in 20% water in MeOH (6 mL). Aqueous NaOH (1N, 3.5 mL) was added, and the reaction mixture was heated to 50 °C for 16 h. The solvents were removed in vacuo and the residue dissolved in water (15 mL). The pH of the solution was adjusted to ~6–7 with 1N aqueous HCl and extracted with EtOAc (100 mL). The organic layer was separated, dried over anhydrous MgSO₄, and the solvent was removed in vacuo to give the compound **16** (0.42 g, 87%). MS *m/z*: 388 (M + H). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 7.32 (d, *J* = 8.3 Hz, 2H), 7.22 (d, *J* = 8.3 Hz, 2H), 7.11 (d, *J* = 8.5 Hz, 2H), 6.84 (d, *J* = 8.5 Hz, 2H), 3.82–3.90 (m, 3H), 3.72 (dd, *J* = 6.7, 9.5 Hz, 1H), 3.00–3.09 (m, 1H), 2.74–2.90 (m, 2H), 2.36 (ddd, *J* = 5.2, 6.8, 12.0 Hz, 1H), 2.14–2.30 (m, 3H), 1.91 (dd, *J* = 8.3, 12.0 Hz, 1H), 1.52–1.75 (m, 5H). HPLC-1: 98.2%, *t*_R = 20.2 min.**

(*R*)-2-[4-(4-Chlorophenoxy)phenoxy]methyl]pyrrolidine Hydrochloride (17**). Compound **41** was prepared from **30** analogous to the procedure described for **40** as a yellow oil (0.37 g). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 7.38 (d, *J* = 8.8 Hz, 2H), 7.00 (s, 4H), 6.93 (d, *J* = 8.8 Hz, 2H), 4.02 (br s, 2H), 3.83–3.93 (m, 1H), 3.27 (br s, 2H), 1.86–2.02 (m, 3H), 1.76–1.84 (m, 1H), 1.40 (s, 9H).**

A solution of 4N HCl in *p*-dioxane (5 mL) was added to **41** (0.36 g, 0.89 mmol). The resulting mixture was stirred at rt for 2 h. The solvent was removed in vacuo. The crude product was triturated with Et₂O and dried in vacuo to afford the compound **17** (0.19 g, 70%) as a light-yellow solid. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 1.75 (m, 1H), 1.87–2.01 (m, 2H), 2.12 (m, 1H), 3.21 (m, 2H), 3.89 (m, 1H), 4.17 (m, 1H), 4.24 (dd, *J* = 4.0, 10.8 Hz, 1H), 6.94 (d, *J* = 9.2 Hz, 2H), 7.05 (s, 4H), 7.40 (d, *J* = 8.8 Hz, 2H). LCMS *m/z*: 304.2 (M + 1). HPLC-1: (99.1%), *t*_R = 19.2 min. Anal. calcd (C₁₇H₁₈ClNO₂·HCl): C, H, N.

3-[(*R*)-2-[4-(4-Chlorophenoxy)phenoxy]methyl]pyrrolidin-1-yl]-propionic Acid Hydrochloride (18**). A solution of **46** (0.62 g, 1.59 mmol) in 1:1 conc. HCl/*p*-dioxane (26 mL) was stirred at 60 °C for 4 h. The solvent was removed in vacuo. The crude product was triturated with ether and the solvent was removed in vacuo to afford the title compound **18** (0.40 g, 68%) as a white solid. MS *m/z*: 374.5 (M – 1). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 1.83 (m, 1H), 2.00 (m, 2H), 2.24 (m, 1H), 2.86 (m, 2H), 3.18 (m, 1H), 3.38 (m, 1H), 3.63 (m, 2H), 3.96 (m, 1H), 4.28–4.37 (m, 2H), 6.94 (d, *J* = 9.2 Hz, 2H), 7.06 (s, 4H), 7.39 (d, *J* = 8.8 Hz, 2H). HPLC-1: 97.7%, *t*_R = 19.5 min.**

4-[(*R*)-2-[4-(4-Chlorophenoxy)phenoxy]methyl]pyrrolidin-1-yl]-l-butyric Acid Hydrochloride (19**). A solution of the methyl ester **47** (0.8 g, 1.99 mmol) in conc. HCl (18 mL) and *p*-dioxane (18 mL) was stirred at 60 °C for 4 h. The solvent was removed in vacuo to provide an oil. The oil was triturated with Et₂O (10 mL), and the resulting solid was filtered to yield the title compound **19** (0.72 g, 93%) as a white solid. MS *m/z*: 388 (M – 1). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 1.82–2.03 (m, 5H), 2.24 (m, 1H), 2.38 (t, *J* = 7.2 Hz, 2H), 3.15 (m, 2H), 3.49 (m, 1H), 3.64 (m, 1H), 3.90 (m, 1H), 4.28–4.34 (m, 2H), 6.94 (d, *J* = 9.2 Hz, 2H), 7.06 (s, 4H), 7.40 (d, *J* = 8.8 Hz, 2H). HPLC-2: 98.1%, *t*_R = 18.5 min.**

4-[(*S*)-2-[4-(4-Chlorophenoxy)phenoxy]methyl]pyrrolidin-1-yl]-butyric Acid Hydrochloride (20**). To a solution of **48** (2.2 g,**

5.45 mmol) in MeOH (22 mL) was added 2N aqueous NaOH (7.2 mL, 14.5 mmol). The resulting solution was stirred at rt overnight. The solvent was removed in vacuo. The crude oil was dissolved in water (45 mL), and the pH was adjusted to 7 with 2N HCl solution. The crude residue was extracted into EtOAc (3 × 250 mL). The combined organic portions were washed with brine, dried over anhydrous Na₂SO₄, filtered, and the solvent was removed in vacuo. The crude product was purified by silica gel flash chromatography using (0–20% MeOH/CH₂Cl₂, gradient) to give the free base of the title compound as a yellow oil. To the oil was added 2 M HCl in Et₂O (35 mL). The resulting mixture was stirred at rt for 2 h. After removal of the supernatant solvent, the remaining white solid was triturated in Et₂O (50 mL) for 1 h. The slurry was filtered, washed with Et₂O (3 × 25 mL), and dried at 45 °C for 48 h to afford the title compound **20** (0.84 g, 36%) as a white solid. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 1.82–2.03 (m, 5H), 2.23 (m, 1H), 2.38 (t, *J* = 7.2 Hz, 2H), 3.14 (m, 2H), 3.48 (m, 1H), 3.63 (m, 1H), 3.90 (m, 1H), 4.28–4.37 (m, 2H), 6.94 (d, *J* = 8.8 Hz, 2H), 7.06 (s, 4H), 7.40 (d, *J* = 8.8 Hz, 2H), 10.55 (br, 1H), 12.30 (br, 1H). HPLC-1: 99.5%, *t*_R = 19.5 min. Anal. calcd (C₂₁H₂₄ClNO₄·HCl): C, H, N.

4-(4-Phenoxyphenyl)piperazine-1-carboxylic Acid *tert*-Butyl Ester (25**). nBu₃P (0.7 g, 3.0 mmol), Pd₂(dba)₃ (0.36 g, 0.3 mmol), and NaOtBu (1.26 g, 13.1 mmol) were added to a solution of **23** (3.9 g, 13.1 mmol) and 1-BOC-piperazine (2.5 g, 13.1 mmol) in toluene (20 mL). The reaction mixture was stirred at rt overnight and then filtered. The solvent was removed in vacuo, and the residue was purified on silica gel (70 g) using 1% MeOH in CH₂Cl₂ to yield the title compound **25** (3.5 g, 75.5%) as a light-yellow solid. MS *m/z*: 355 (M + H). LCMS 99%. ¹H NMR (400 MHz, CDCl₃) δ: 7.57 (d, *J* = 8.9 Hz, 2H), 6.89–6.98 (m, 4H), 6.71 (d, *J* = 8.9 Hz, 2H), 3.55–3.62 (m, 4H), 3.05–3.11 (m, 4H), 1.49 (s, 9H).**

4-[4-(4-Iodophenoxy)phenyl]piperazine-1-carboxylic Acid *tert*-Butyl Ester (26**). To a solution of **24** (4.22 g, 10 mmol), 1-BOC-piperazine (1.86 g, 10 mmol) were added Pd₂(dba)₃ (0.270 g, 0.295 mmol), nBu₃P (0.58 g, 2.8 mmol), and NaOtBu (0.95 g, 10 mmol). The reaction mixture was stirred at rt overnight and then filtered. The solvent was removed in vacuo, and the residue was purified on silica gel (100 g) using 1% MeOH in CH₂Cl₂ to yield the title compound **26** (1.2 g, 25%) as a yellow solid. MS *m/z*: 481 (M + H), LCMS 97%. ¹H NMR (400 MHz, CDCl₃) δ: 7.57 (d, *J* = 8.9 Hz, 2H), 6.93 (dd, *J* = 16.0, 9.4 Hz, 4H), 6.71 (d, *J* = 8.7 Hz, 2H), 3.55–3.61 (m, 4H), 3.05–3.12 (m, 4H), 1.49 (s, 9H).**

(*S*)-2-(Toluene-4-sulfonyloxymethyl)pyrrolidine-1-carboxylic Acid *tert*-Butyl Ester (29**). To a solution of (*S*)-*N*-BOC-prolinol (22 g, 110 mmol) in pyridine (56 mL) at 0 °C was added a solution of *p*-toluenesulfonyl chloride (22.9 g, 120 mmol) in pyridine (56 mL) portionwise over 5 min. The pale-yellow reaction mixture was stirred at 0 °C for 2 h and then at rt overnight. The solvent was removed in vacuo. The crude oil was extracted into EtOAc (400 mL) and sequentially washed with 0.5 M HCl (100 mL), saturated aqueous NaHCO₃ (100 mL), and brine (100 mL). The combined organic layer was dried over anhydrous Na₂SO₄, filtered, and the solvent was removed in vacuo to give compound **27** (39 g, 99%) as a yellow oil. MS *m/z*: 356 (M + H). ¹H NMR (400 MHz, CDCl₃) δ: 7.79 (d, *J* = 8.1 Hz, 2H), 7.31–7.40 (m, 2H), 4.06–4.18 (m, 2H), 3.90 (br s, 1H), 3.30 (d, *J* = 6.3 Hz, 2H), 2.45 (s, 3H), 1.74–2.00 (m, 4H), 1.39 (d, *J* = 16.4 Hz, 9H).**

(*R*)-2-(Toluene-4-sulfonyloxymethyl)pyrrolidine-1-carboxylic Acid *tert*-Butyl Ester (30**). This compound was prepared analogously to the preparation of **29** from (*R*)-BOC-prolinol **28** (500 mg, 2.48 mmol) and tosyl chloride (565 mg, 2.96 mmol) in pyridine (2.5 mL) to provide, after aqueous workup and concentration of organic solvent in vacuo, the tosylate **30** (800 mg, 91%) as a thick oil. MS *m/z*: 378 (M + Na). ¹H NMR (400 MHz, CDCl₃) δ: 1.38 (m, 9H), 1.79 (m, 2H), 1.93 (m, 2H), 2.44 (s, 3H),**

3.26–3.32 (m, 3H), 3.88–3.97 (m, 2H), 4.07–4.14 (m, 2H), 7.34 (br, 2H), 7.77 (d, 2H, $J = 8.0$ Hz).

(4-Chlorophenyl)(4-methoxyphenyl)methanone (32a). Prepared according to the procedure for **32b** from 4-chlorobenzoyl chloride. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ : 7.80 (d, $J = 8.7$ Hz, 2H), 7.71 (d, $J = 8.5$ Hz, 2H), 7.45 (d, $J = 8.5$ Hz, 2H), 6.97 (d, $J = 8.7$ Hz, 2H), 3.89 (s, 3H).

(4-Iodophenyl)(4-methoxyphenyl)methanone (32b). Cold nitrobenzene (45 mL) was treated portionwise with AlCl_3 (13.5 g, 101 mmol). 4-Iodobenzoyl chloride **31b** (25 g, 94 mmol) in nitrobenzene (20 mL) was added, keeping the reaction temperature at $< 10^\circ\text{C}$. The mixture was stirred at 0°C for 10 min, and anisole (9.5 g, 88 mmol) was added dropwise to maintain temperature at $< 10^\circ\text{C}$. The reaction was allowed to rt over 16 h and then poured into ice water (750 mL). The resulting solids were removed by filtration, and the filtrate was extracted with CH_2Cl_2 (2 L). The organic materials were washed with NaHCO_3 (2×150 mL), dried over anhydrous MgSO_4 , and the solvent was removed in vacuo. The residue was triturated with cyclohexane (200 mL) to obtain **32b** as a solid (26.75 g, 90%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ : 7.84 (d, $J = 8.3$ Hz, 2H), 7.80 (d, $J = 8.9$ Hz, 2H), 7.48 (d, $J = 8.3$ Hz, 2H), 6.96 (d, $J = 8.9$ Hz, 2H), 3.89 (s, 3H).

4-(4-Chlorobenzyl)methoxybenzene (33b). A solution of **32b** (9.8 g, 39.7 mmol) in TFA (30 mL) was cooled to 0°C . To this solution was added triethylsilane (19.22 g, 164.7 mmol), and the reaction was allowed to rt for 16 h. The reaction was poured into water (150 mL) and was adjusted to pH 6–7 with 1N NaOH. The solution was extracted with EtOAc (400 mL), and the organic layer was washed with water (200 mL) and then dried over anhydrous MgSO_4 . The solvent was removed in vacuo to give the title product **33b** (9.2 g, 99%), which was used without further purification for the next step. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ : 7.25 (d, $J = 3.9$ Hz, 2H), 7.09 (d, $J = 8.4$ Hz, 2H), 7.07 (d, $J = 8.4$ Hz, 2H), 6.83 (d, $J = 8.6$ Hz, 2H), 3.88 (s, 2H), 3.78 (s, 3H).

4-(4-Chlorobenzyl)phenol (34a). Prepared according to the procedure for **34b** starting with 4-(4-chlorobenzyl)methoxybenzene. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ : 7.30 (d, $J = 3.4$ Hz, 2H), 7.13 (d, $J = 8.3$ Hz, 2H), 7.06 (d, $J = 8.5$ Hz, 2H), 6.80 (d, $J = 8.5$ Hz, 2H), 4.74 (s, 1H), 3.91 (s, 2H).

4-(4-Iodobenzyl)phenol (34b). To a solution of **32b** (26.7 g, 79 mmol) in TFA (90 mL) at 0°C , triethylsilane (29 mL, 181 mmol) was added slowly. The reaction was stirred at rt for 16 h. The volatiles were removed in vacuo. The residue was dissolved in EtOAc (200 mL), washed with NaHCO_3 (2×300 mL) and 6N HCl (2×50 mL), and then dried over anhydrous MgSO_4 . Removal of solvent in vacuo gave an oil, which was used without further purification. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ : 7.59 (d, $J = 8.2$ Hz, 2H), 7.06 (d, $J = 8.6$ Hz, 2H), 6.92 (d, $J = 8.2$ Hz, 2H), 6.82 (d, $J = 8.6$ Hz, 2H), 3.85 (s, 2H), 3.78 (s, 3H). To a -78°C solution of crude **33b** in CH_2Cl_2 (150 mL) was added BBr_3 (1 M in CH_2Cl_2 , 158 mL, 158 mmol) while maintaining the reaction temperature below -65°C . The reaction was allowed to come to rt, then quenched by pouring over ice water (1 L). The aqueous material was extracted with CH_2Cl_2 (2×100 mL). The organic layer was washed with NaHCO_3 (2×2000 mL) and brine (100 mL) and then dried over anhydrous MgSO_4 . The solvent was removed in vacuo, and the residue triturated with hexanes to obtain the product **34b** (22.3 g, 91%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ : 7.59 (d, $J = 8.2$ Hz, 2H), 7.01 (d, $J = 8.3$ Hz, 2H), 6.91 (d, $J = 8.2$ Hz, 2H), 6.75 (d, $J = 8.5$ Hz, 2H), 4.68 (s, 1H), 3.84 (s, 2H).

4-(4-Thiophen-3-yl-benzyl)phenol (35). A mixture of **34b** (12.4 g, 40 mmol), 3-thienylboronic acid (6.15 g, 48 mmol), Pd/C (2.12 g, 2 mmol), and K_2CO_3 (16.6 g, 120 mmol) was suspended in IPA/ H_2O (5:1, 240 mL) under an argon purge. The stirred mixture was heated at 85°C for 16 h. After cooling to rt, the reaction was passed through a pad of celite and concentrated in vacuo. The residue was triturated with water and the solids were

isolated by filtration to give the title product **35** (10.6 g, 99%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ : 7.51 (d, $J = 8.1$ Hz, 2H), 7.39–7.43 (m, 1H), 7.36 (d, $J = 2.2$ Hz, 2H), 7.19 (d, $J = 8.1$ Hz, 2H), 7.07 (d, $J = 8.3$ Hz, 2H), 6.76 (d, $J = 8.5$ Hz, 2H), 4.68 (s, 1H), 3.93 (s, 2H).

(R)-2-[4-(4-Chlorobenzyl)phenoxyethyl]pyrrolidine (37). Synthesis was carried out analogous to procedure used for the preparation of **15** starting with **34a**. BOC protected intermediate, (R)-2-[4-(4-chlorobenzyl)-phenoxyethyl]-pyrrolidine-1-carboxylic acid *tert*-butyl ester, was purified by flash column chromatography (15% EtOAc/hexanes, isocratic). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ : 7.23 (d, $J = 8.2$ Hz, 2H), 7.09 (d, $J = 8.3$ Hz, 2H), 7.05 (d, $J = 8.1$ Hz, 2H), 6.85 (d, $J = 6.5$ Hz, 2H), 4.04–4.21 (m, 2H), 3.87 (s, 2H), 3.68–3.81 (m, 1H), 3.27–3.46 (m, 2H), 1.78–2.03 (m, 4H), 1.46 (s, 9H).

4-[(R)-2-[4-(4-Chlorobenzyl)phenoxyethyl]pyrrolidin-1-yl]-butyric Acid Methyl Ester (38). A solution of **37** (1.6 g, 4 mmol) was treated with 4 N HCl in *p*-dioxane (25 mL) for 6 h at rt, the solvent was removed in vacuo, and residue dried under high vacuum for 4 h to provide solid (1.4 g) that was used as such for the next step. To solid (1.4 g, 4 mmol) and methyl-4-bromobutyrate (0.9 g, 5 mmol) in DMF (15 mL) was added K_2CO_3 (1.1 g, 8 mmol). The reaction mixture was stirred at rt for 16 h. The solvent was removed in vacuo, and the residue was partitioned between water (35 mL) and EtOAc (130 mL). The organic layer was dried over anhydrous MgSO_4 , the solvent was removed in vacuo, and residue was purified by flash chromatography (50% EtOAc/hexanes, isocratic) to give the ester **38** (0.57 g, 35%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ : 7.25 (d, $J = 6.6$ Hz, 2H), 7.09 (d, $J = 8.3$ Hz, 2H), 7.05 (d, $J = 8.6$ Hz, 2H), 6.82 (d, $J = 8.6$ Hz, 2H), 3.86–3.92 (m, 3H), 3.74 (dd, $J = 6.9, 9.1$ Hz, 1H), 3.64 (s, 3H), 3.15 (d, $J = 4.0$ Hz, 1H), 2.79–2.92 (m, 2H), 2.30–2.47 (m, 3H), 2.24 (q, $J = 8.3$ Hz, 1H), 1.96 (d, $J = 11.8$ Hz, 1H), 1.63–1.88 (m, 5H).

(S)-2-[4-(4-Iodophenoxyethyl)pyrrolidine-1-carboxylic Acid *tert*-Butyl Ester (40). To a solution of 4-iodophenol (24.2 g, 110 mmol) in DMF (200 mL) was added KOtBu (12.3 g, 110 mmol). The cloudy solution was stirred at rt for 30 min before a solution of **29** (35.9 g, 100 mmol) in DMF (200 mL) was added. The reaction was heated to 45°C for 16 h and then increased to 55°C for 8 h. The solvent was removed in vacuo, and the residue was partitioned between hexanes (500 mL) and water (250 mL). The organic layer was washed with 0.2 N NaOH (2×200 mL) and H_2O (250 mL). The solvent was reduced to 20% and allowed to stand at rt for 1 h. The supernatant liquid was decanted, and the remaining solids were washed with hexane (50 mL). The solid was dried under high vacuum at rt to give the title compound **40** (37.0 g, 92%). $^1\text{H NMR}$ (400 MHz, $\text{DMSO}-d_6$) δ : 7.58 (d, $J = 8.7$ Hz, 2H), 6.82 (d, $J = 8.8$ Hz, 2H), 3.95–4.07 (m, 2H), 3.82–3.93 (m, 1H), 3.26 (br. s., 2H), 1.71–2.03 (m, 4H), 1.39 (s, 9H).

(S)-2-[4-(4-Chlorophenoxy)phenoxyethyl]pyrrolidine-1-carboxylic Acid *tert*-Butyl Ester (42). A mixture of **40** (11.8 g, 29.3 mmol), 4-chlorophenol (11.3 g, 87.9 mmol), copper(I) iodide (1.4 g, 7.3 mmol), cesium carbonate (38.2 g, 117.2 mmol), and *N,N*-dimethylglycine (1.5 g, 14.6 mmol) in *p*-dioxane (200 mL) was heated to reflux under an N_2 atmosphere for 16 h. The reaction mixture was concentrated in vacuo, and residue was partitioned between hexanes (500 mL) and water (250 mL). The organic layer was washed with 0.2N NaOH (2×250 mL) and water (250 mL) and then concentrated in vacuo to give a crude oil. The oil was purified by silica gel flash chromatography (10% acetone/hexanes, isocratic) to give the title compound **42** (6.6 g, 56%). MS m/z : 404 (M + H). $^1\text{H NMR}$ (400 MHz, $\text{DMSO}-d_6$) δ : 7.38 (d, $J = 8.8$ Hz, 2H), 7.00 (s, 4H), 6.93 (d, $J = 8.8$ Hz, 2H), 3.96–4.09 (m, 2H), 3.83–3.94 (m, 1H), 3.27 (br. s., 2H), 1.86–2.04 (m, 3H), 1.74–1.85 (m, 1H), 1.40 (s, 9H).

(S)-2-[4-(4-Chlorophenoxy)phenoxyethyl]pyrrolidine (43). Compound **42** (6.6 g, 16.4 mmol) was stirred in 4 N HCl in *p*-dioxane at rt for 2 h. The solvent was removed in vacuo, and

the residue was partitioned between hexanes (200 mL) and water (200 mL). The organic layer was extracted with 1 N HCl (25 mL). The aqueous layers were combined, neutralized with Na₂CO₃ (solution), and extracted with hexanes (2 × 100 mL). The solvent was removed in vacuo at 40 °C until the solution became cloudy. The mixture was allowed to stand at rt and then refrigerated for 1 h. The supernatant liquid was decanted, and the resulting solids were dried under vacuum at rt to give the title compound **43** (4.1 g, 83%). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 7.38 (d, *J* = 8.7 Hz, 2H), 6.87–7.05 (m, 6H), 3.73–3.84 (m, 2H), 3.26–3.43 (m, 1H), 2.72–2.88 (m, 2H), 2.30–2.46 (m, 1H), 1.78–1.92 (m, 1H), 1.54–1.76 (m, 2H), 1.35–1.52 (m, 1H).

Methyl 3-[(2*R*)-2-[(4-(4-Chlorophenoxy)phenoxy)methyl]-1-pyrrolidinyl]propanoate (46). To a solution of **17** (0.8 g, 2.63 mmol) in CH₂Cl₂ (7.6 mL) was added Et₃N (0.78 mL, 5.60 mmol) and methyl 3-bromopropionate (0.32 mL, 2.93 mmol). The resulting solution was stirred at 30 °C overnight. The reaction mixture was poured into water/CH₂Cl₂ (50 mL/50 mL). The crude residue was extracted into CH₂Cl₂ (50 mL). The organic portion was washed with water (50 mL), washed with brine (50 mL), dried over anhydrous Na₂SO₄, filtered, and concentrated in vacuo. The crude product was purified by silica gel flash chromatography (0–50% EtOAc/hexanes, gradient) to give the title compound **46** (0.79 g, 77%). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 7.34–7.42 (m, 2H), 6.90–7.04 (m, 6H), 3.90 (dd, *J* = 5.1, 9.4 Hz, 1H), 3.74 (dd, *J* = 6.5, 9.5 Hz, 1H), 3.57 (s, 3H), 3.15 (d, *J* = 12.2 Hz, 1H), 2.98–3.06 (m, 1H), 2.83 (dd, *J* = 5.9, 8.1 Hz, 1H), 2.56–2.66 (m, 1H), 2.38–2.48 (m, 2H), 2.23 (d, *J* = 7.9 Hz, 1H), 1.85–1.98 (m, 1H), 1.65–1.76 (m, 2H), 1.59 (s, 1H).

4-[(*R*)-2-(4-(4-Chlorophenoxy)phenoxy)methyl]pyrrolidin-1-yl]-l-butyric Acid Methyl Ester (47). To a solution of **17** (0.8 g, 2.63 mmol) in CH₂Cl₂ (7.6 mL) was added Et₃N (0.78 mL, 5.60 mmol) and methyl 4-bromobutyrate (0.35 mL, 3.04 mmol). The resulting solution was stirred at 30 °C overnight. The reaction mixture was poured into water/CH₂Cl₂ (50 mL/50 mL). The crude residue was extracted into CH₂Cl₂. The organic portion was washed with water (50 mL), brine (50 mL), dried over anhydrous Na₂SO₄, filtered, and concentrated in vacuo. The crude product was purified by silica gel flash chromatography using (0–50% EtOAc/hexanes, gradient) to give the methyl ester **47** (0.60 g, 57%). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 7.38 (d, *J* = 8.8 Hz, 2H), 6.89–7.03 (m, 6H), 3.87 (dd, *J* = 5.0, 9.5 Hz, 1H), 3.73 (dd, *J* = 6.6, 9.4 Hz, 1H), 3.55 (s, 3H), 2.98–3.08 (m, *J* = 3.6 Hz, 1H), 2.81–2.89 (m, *J* = 12.0 Hz, 1H), 2.73–2.81 (m, *J* = 1.9 Hz, 1H), 2.29–2.40 (m, 3H), 2.17 (q, *J* = 7.9 Hz, 1H), 1.85–1.98 (m, 1H), 1.63–1.77 (m, 4H), 1.53–1.63 (m, 1H).

4-[(*S*)-2-[4-(4-Chlorophenoxy)phenoxy)methyl]pyrrolidin-1-yl]-butyric Acid Methyl Ester (48). To a solution of **43** (4.0 g, 13.2 mmol) in DMF (50 mL) was added K₂CO₃ (3.7 g, 26.5 mmol) and methyl 4-bromobutyrate (3.6 g, 19.8 mmol). The resulting suspension was stirred at 55 °C for 16 h. The solvent was removed in vacuo, and the residue was partitioned between hexanes (250 mL) and water (150 mL). The organic layer was washed with water (150 mL), dried over anhydrous Na₂SO₄, filtered, and concentrated in vacuo. The crude product was purified by silica gel flash chromatography (1–2% iPrOH/EtOAc + 0.25% TEA, gradient) to give the title compound **48** (4.1 g, 77%). MS *m/z*: 404 (M + H). ¹H NMR (400 MHz, DMSO-*d*₆) δ: 7.38 (d, *J* = 8.8 Hz, 2H), 6.89–7.03 (m, 6H), 3.87 (dd, *J* = 5.1, 9.4 Hz, 1H), 3.73 (dd, *J* = 6.7, 9.3 Hz, 1H), 3.55 (s, 3H), 3.00–3.08 (m, 1H), 2.73–2.90 (m, 2H), 2.30–2.40 (m, 3H), 2.17 (q, *J* = 8.2 Hz, 1H), 1.85–2.01 (m, 1H), 1.64–1.76 (m, 4H), 1.60 (q, *J* = 6.0 Hz, 1H).

Biological Assays. In Vitro Enzyme Assay. LTA₄ substrate was prepared from the methyl ester of LTA₄ (BioMol, Plymouth Meeting, PA, or Cayman Chemicals, Ann Arbor, MI) by treatment under nitrogen with 100 mol equiv of NaOH in an acetone:H₂O (4:1) solution at rt for 40 min. Stock solutions of

LTA₄ were kept frozen at –80 °C for a maximum of one week prior to use.

Recombinant human LTA₄H (200 nM) was incubated with various concentrations of test compound for 10 min at rt in assay buffer (0.1 M Tris-HCl, 0.1 M NaCl, 5 mg/mL fatty acid free BSA, 10% DMSO, pH 8.0). Immediately before the assay, LTA₄ was diluted to a concentration of 10 μM in assay buffer (without DMSO) and added to the reaction mixture to a final concentration of 2 μM to initiate the enzyme reaction. Samples were incubated for 2 min at rt, followed by the addition of 2 volumes of chilled quenching buffer (CH₃CN with 1% HOAc and 225 nM LTB₄-d₄ (BioMol)). The samples were then stored at 4 °C overnight to complete protein precipitation and centrifuged for 15 min at 1800g. LTB₄ formed was measured in the supernatant by LC-MS/MS using LTB₄-d₄ as an internal standard and an external LTB₄ standard (BioMol) for a calibration curve. Briefly, the analyte was separated from LTB₄ isomers formed by spontaneous hydrolysis of LTA₄ using isocratic elution on an HPLC system (Waters, Milford, MA) and analyzed on a tandem quadrupole mass spectrometer (Waters Micromass Quattro Premier). MRM transitions followed on 2 channels were 335.2 → 195.3 (LTB₄) and 339.2 → 197.3 (LTB₄-d₄). On the basis of the measured amounts of LTB₄ formed at each inhibitor concentration, a dose–response curve was fitted to the data using a sigmoidal 4-parameter function (XLfit, model 205) and an IC₅₀ value was calculated.

Human Whole Blood LTB₄ Assay. Human blood (45 mL) was collected in heparin-containing Vacutainer tubes (Greiner-Bio One) with informed consent. Individual experiments were performed with blood from a single subject. For each sample, 200 μL of blood were dispensed into a prewarmed 96-well plate and 188 μL of RPMI-1640 medium (Invitrogen) containing 20 μg/mL Indomethacin (Sigma, St. Louis, MO) was added. Then 4 μL of a series of compound dilutions (final DMSO concentration of 1%) were added in triplicate, followed by a 15 min incubation at 37 °C with gentle shaking. After that, blood samples were stimulated by adding 8 μL of ionomycin (from *Streptomyces globatus*, Calbiochem) to a final concentration of 36 μM. After another incubation at 37 °C for 30 min, samples were centrifuged at 4 °C for 5 min at 1800g. LTB₄ concentrations in supernatants were determined using a commercially available enzyme-linked immunosorbent assay (R&D Systems, Minneapolis, MN) according to the manufacturer's instructions. On the basis of the measured amounts of LTB₄ formed at each inhibitor concentration, a dose–response curve was fitted to the data using a sigmoidal three-parameter function (XLfit, model 203 with Hill slope fixed to –1) and an IC₅₀ value was calculated. Each assay plate contained eight positive controls (no compound) and eight negative controls (no ionomycin).

X-ray Crystallography. With PEG8000 as precipitant, ytterbium as an additive, and the use of microseeding to control nucleation rate, well-diffracting crystals of the enzyme in complex with the micromolar inhibitor Bestatin were obtained routinely. These crystals were subsequently used to soak in various high-affinity lead compounds of interest, replacing Bestatin. Using this strategy, a crystal structure of LTA₄H in complex with compounds **2**, **5**, **6**, **7**, **8**, **9**, **13**, **14**, **15**, and **20** were determined at 1.67, 2.16, 1.8, 2.0, 2.05, 2.4, 1.90, 1.6, 1.63, and 2.5 Å resolution, respectively. The carboxylate group of compound **20** chelates the catalytic Zn ion, its oxygens replacing two water oxygens in almost identical positions in the structure with compound **14**. The two phenyl rings of both compounds bind in a very similar mode to the end of the bend, hydrophobic substrate binding pocket, which has three conserved waters tightly bound.

Oral Dosing of Compounds to Rats. The compound was formulated in 10% Solutol HS15 (BASF AG, Ludwigshafen, Germany), 30 mM NaH₂PO₄, pH 7.9. The same solution without the compound was used for vehicle control experiments. The

solutions were given by orogastric gavage to male Sprague–Dawley rats in a dosing volume of 10 mL/kg (1% of body weight) using stainless steel feeding needles of 0.9 mm × 38 mm (B & K Universal AB, Sollentuna, Sweden) and 1 mL Omnifix syringes (B. Braun, Melsungen, Germany). The rats were fasting in the single dose time course experiments and nonfasting in all the other experiments.

Isolation of Whole Blood from Dosed Animals. Rats were put in a special cylinder to hold them still. The tail vein was punctured with 25G × 5/8 in. needle and the blood collected into a heparin tube. Dosing in dog and monkey was carried out at CROs sites.

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Supporting Information Available: Elaboration of fragments, hydrophilic linkers for which bound-structures are shown, hydrolase-bound fragments containing hydrophilic linkers, plot of BEI vs. SEI, preparation of recombinant human LTA₄ hydrolase, protein crystallization, preparation of compound-soaked crystals and crystal freezing, and data collection, structure solution, and refinement. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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